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DICHIARAZIONE DI UNICITA'

Si dichiara che:

Lo spettrometro di massa a filtro Quadrupolare accoppiato ad un sistema FT MS basato sulla nuova tecnologia Orbitrap, denominato Q-EXACTIVE Plus presenta le seguenti caratteristiche che lo rendono unico nel mercato della Spettrometria di Massa.

Spettrometro di Massa a Trasformata di Fourier (FTMS) funzionate in base ad un nuovo concetto brevettato da Thermo Fisher, che non utilizza campi magnetici (generati da magneti a superconduzione con conseguenti sistemi criogenici), radiofrequenze o misura di tempi di volo (TOF) per separare gli ioni di diverso valore di massa/carica come negli spettrometri attualmente in commercio, ma utilizza un (semplice) campo elettrostatico applicato ad un elettrodo centrale intorno al quale ruotano in modo Radiale ed Assiale gli ioni stessi.

La determinazione del rapporto Massa su Carica (m/z) degli ioni avviene misurando nel tempo la corrente generata dalla movimento assiale degli ioni medesimi.

La Trasformata di Fourier viene utilizzata per convertire il segnale acquisito nel tempo in quello delle diverse frequenze ed intensità che lo compongono da cui viene calcolata la massa e la quantità degli ioni.

La risoluzione dello spettro di massa ottenuto dipende dalla durata del tempo di osservazione senza perdita di intensità del segnale al suo aumentare.

Lo spettrometro di massa Q Exactive Plus e' in grado di garantire una Risoluzione massima di 140.000 FWHM misurata a m/z 200

La tecnologia Orbitrap è protetta dai seguenti brevetti:

•	Orbitrap	Patent Number	US 6,872,938 B2
•	Orbitrap	Patent Number	US 5,886,346

Thermo Fisher Scientific	Sede Legale: Strada	+39 02 95059 1	www.thermofisher.com
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Società con socio unico - Direzione e Coordinamento Thermo Fisher Scientific Holdings



Inoltre lo Spettrometro di Massa Q Exactive possiede le seguenti caratteristiche uniche aggiuntive:

Possibilità diverse di frammentazione:

• HCD (Higher Energy Collision Dissociation) nella cella quadrupolare dell'Orbitrap con generazione di uno spettro di frammentazione.

Possibilità di isolamento dello ione precursore in alta risoluzione.

- HRI con isolamento pari a 0.4 amu.
- Data Dependent ScanTM Possibilità di acquisire spettri MS-Full Scan e MS/MS.

Si dichiara inoltre che l'unica società ufficialmente riconosciuta dalla casa madre autorizzata a rivendere la propria strumentazione, nel dettaglio gli spettrometri di massa a filtro quadrupolare accoppiati ad un sistema FT-MS basati sulla nuova tecnologia Orbitrap e denominati Q Exactive Plus, è unicamente ed esclusivamente Thermo Fisher Scientific S.p.a. su tutto il territorio nazionale.

Thermo Fisher Scientific Sip.A.



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Makarov et al.

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(54) MASS SPECTROMETRY METHOD AND APPARATUS

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- Nov. 7, 2001 (GB) 0126764
- 250/288

(56) References Cited

U.S. PATENT DOCUMENTS

5,179,278	Α		1/1993	Douglas	
5,420,425	А		5/1995	Bier et al.	
5,569,917	Α		10/1996	Buttrill, Jr. et al.	
5,572,022	Α		11/1996	Schwartz et al.	
5,729,014	Α	*	3/1998	Mordehai et al	250/282
5,880,466	Α		3/1999	Benner	
5,886,346	Α		3/1999	Makarov	
6,011,259	Α		1/2000	Whitehouse et al.	
6,020,586	Α		2/2000	Dresch et al.	
6,627,883	B2	*	9/2003	Wang et al.	250/292

FOREIGN PATENT DOCUMENTS

EP	0113207 A2	7/1984
EP	0529885 A1	3/1993
WO	WO98/49710	11/1998
WO	WO99/30350	6/1999

OTHER PUBLICATIONS

Baba et al., "Mass Spectrometer", Pub. No. US 2003/ 0222214 A1, publication date: Dec. 4, 2003.*

Michael et al., "An Ion Trap Storage/Time-of-Flight Mass Spectrometer", Rev. Sci. Instrum. 63 (10), Oct. 1992, pp. 4277-4284.

Wollnik, "Energy–Isochronous Time–of–Flight Mass Analyzers", International Journal of Mass Spectrometry and Ion Processes 131 (1994), pp. 387–407.

Senko et al., "External Accumulation of Ions for Enhanced Electrospray Ionization Fourier Transform Ion Cyclotron Resonance Mass Spectrometry", American Society for Mass Spectrometry, 1997, pp. 970–976.

Piyadasa et al., "A High Resolving Power Multiple Reflection Matrix-assisted Laser Desorption/Ionization Time-of-Flight Mass Spectrometer", Rapid Communications in Mass Spectrometery, vol. 13, 1999, pp. 620–624.

Belov et al., "Design and Performance of an ESI Interface for Selective External Ion Accumulation Coupled to a Fourier Transform Ion Cyclotron Mass Spectrometer", Analytical Chemistry, vol. 73, 2001, pp. 253–261.

Lisheng Yang et al., "Confinement of Injected Beam Ions in a Kingdon Trap" Nuclear Instruments and Methods in Physics Research, Section—B: Beam Interactions with Materials and Atoms, North–Holland Publishing Company, Amsterdam, NL, vol. B56/75, No. Part 2, May 1, 1991, pp. 1185–1187.

Gillig K J et al., "Ion Motion in a Fourier Transform Ion Cyclotron Resonance Wire Ion Guide Cell," International Journal of Mass Spectrometry and Ion Processes, Elsevier Scientific Publishing Co. Amsterdam, NL, vol. 157–15, No. Double, Dec. 20, 1996, pp. 129–147.

* cited by examiner

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(57) ABSTRACT

A mass spectrometer 10 comprises an ion source 12 which generates nebulized ions which enter an ion cooler 20 via an ion source block 16. Ions within a window of m/z of interest are extracted via a quadrupole mass filter 24 and passed to a linear trap 30. Ions are trapped in a potential well in the linear trap 30 and are bunched at the bottom of the potential well adjacent an exit segment 50. Ions are gated out of the linear trap 30 into an electrostatic ion trap 130 and are detected by a secondary electron multiplier 10.

By bunching the ions in the linear trap 30 prior to ejection, and by focussing the ions in time of flight (TOF) upon the entrance of the electrostatic trap 130, the ions arrive at the electrostatic trap 130 as a convolution of short, energetic packets of similar m/z. Such packets are particularly suited to an electrostatic trap because the FWHM of each packet's TOF distribution is less than the period of oscillation of those ions in the electrostatic trap.

50 Claims, 8 Drawing Sheets

















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MASS SPECTROMETRY METHOD AND APPARATUS

FIELD OF INVENTION

This invention relates to a method and an apparatus of mass spectrometry, and in particular to a method and an apparatus for storage and injection of ions into an electrostatic ion trap.

BACKGROUND OF THE INVENTION

Mass spectrometers have been used to analyse a wide range of materials, including organic substances such as pharmaceutical compounds, environmental compounds and biomolecules. They are particularly useful, for example, for DNA and protein sequencing. In such applications, there is an ever increasing desire for high mass accuracy, as well as high resolution of analysis of sample ions by the mass spectrometer, notwithstanding the short time frame of modern separation techniques such as gas chromatography/mass²⁰ spectrometry (GC/MS), liquid chromatography/mass spectrometry (LC/MS) and so forth.

One of the new directions in the field of mass spectrometry is the development of mass analysers where ions are dynamically trapped in an electrostatic field. Broadly, these may be divided into two classes: those that employ frequency analysis by image current detection, as disclosed in U.S. Pat. No. 5,880,466 and U.S. Pat. No. 5,886,346, and those that employ time of flight (TOF) separation and ion $_{30}$ detection by secondary electron conversion, as is disclosed, for example by H. Wollnik, in J. Mass Spectrom. Ion Proc. (1994), vol. 131, at pages 387-407, and by C. Piadyasa et al., in Rapid Commun. Mass Spectrom. (1999), vol. 13 at pages 620-624. Although the trap fields may be ramped at the beginning of the mass scan, they are typically held very stable during the detection, or TOF separation of ions, and so each of the foregoing mass analysers may be regarded as electrostatic traps (ESTs).

Such EST mass analysers can achieve high and even 40 ultra-high mass resolutions (in excess of 100,000), thus allowing more accurate determination of ion masses. However, they all operate using an inherently pulsed technique and as such the task of coupling to any external continuous ion source is a serious problem.

To improve duty cycle and sensitivity, it is possible to use an external collision quadrupole ion trap for ion cooling and storage between injections. This technique has proved particularly successful when combined with other inherently pulsed techniques such as TOF mass analysis as is described 50 by S. Michael et al., in Rev. Sci. Instrum. (1992) vol. 63, pages 4277 to 4284. Here, ions are accumulated in the trap. As suggested in U.S. Pat. No. 5,572,022, it is possible to control the number of ions in the trap to reduce space-charge effects. Once ions have been stored in the trap, they can be 55 pulsed into the TOF mass analyser by applying high voltages to the (normally grounded) end caps of the trap. In U.S. Pat. No. 5,569,917, the ions are given a simultaneous "push" out of the trap and a "suck" from the TOF mass analyser, so as to improve the efficiency of ion injection into the analyser. $_{60}$ The spatially spread ion beam is focussed into a tight pack in the "object" plane of the TOF mass analyser.

Despite these improvements, quadrupole ion traps are still currently a relatively inefficient technique for injecting ions into a mass analyser (down to a few percent), and they also 65 suffer from low space charge capacity due to the limited trap volume.

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One approach that has been taken to address these problems is to employ a different type of collisonal storage device known as a linear trap (LT) or RF multipole trap. U.S. Pat. No. 5,179,278 shows such an arrangement, wherein a two-dimensional multipole RF field is generated. The trap of U.S. Pat. No. 5,179,278 is limited by end lenses. Alternatively, the poles of the trap may be split into sections as is shown in U.S. Pat. No. 5,420,425. Both split poles and end caps can be employed together. The elevated voltages on 10 the end lenses or sections limit the ion movement along the axis whilst the RF voltage provides a quasi-potential well in the radial direction. If ions lose enough energy during the first passes through the multipole, then they may be trapped in it and squeezed towards the axis during further collisions. The number of ions in the trap can be controlled using a short pre-scan, a technique disclosed in the abovereferenced U.S. Pat. No. 5,572,022. Nevertheless, to inject ions from the LT into the next stage of analysis, the voltage is lowered on the exit lens and the ions in the LT are allowed to flow out of the multipole. This flow typically lasts up to hundreds or even thousands of microseconds. These time scales are compatible with the injection times for quadrupole ion traps (as disclosed in the above-referenced U.S. Pat. No. 5,179,278) or for Fourier Transform Ion Cyclotron Resonance (FTICR) as set out by M. Senko et al in JASMS, (1997), volume 8, pages 970-976. The time scales are also suitable for orthogonal acceleration TOF mass spectrometry, see for example U.S. Pat. No. 6,011,259, U.S. Pat. No. 6,020,586, and WO99/30350.

Segmented construction of the poles in the LT may be employed, as set out by M. Belov et al, in Analytical Chemistry (2001) volume 73, pages 253-261, to reduce the injection time down to about 300-400 microseconds. The segmented construction of the LT provides an axial field which causes ions to be displaced towards the exit lens.

Even so, such injection times are too long for an electrostatic trap. This is because ESTs require high ion energies (typically 1–2 keV per charge) to achieve dynamic trapping. If injection takes place over hundreds of microseconds, at such energies the process may last for hundreds of ion reflections. Without any collisional cooling inside the electrostatic trap, ion stability may be compromised.

SUMMARY OF THE INVENTION

It is an object of the present invention to provide a method and an apparatus which alleviates these problems with the prior art. In particular, it is an object to provide a method and an apparatus which provides for adequate storage of ions prior to injection of these ions into an electrostatic trap over a timescale compatible with such a device.

According to a first aspect of the present invention, there is provided a method of injection of sample ions into an electrostatic trap, comprising the steps of: (a) generating a plurality of sample ions to be analysed, each of which has a mass-to-charge ratio m/z; (b) receiving the sample ions through a storage device entrance in an ion storage device having a plurality of storage device poles; (c) supplying a trapping voltage to the storage device so as to trap at least a proportion of the received sample ions within a volume ρ in the storage device, during at least a part of a trapping period, the thus trapped ions each having a kinetic energy E_k such that there is an average kinetic energy \overline{E}_{k} of the ions in the volume ρ during the trapping period or the said part thereof; (d) supplying a release voltage to the storage device so as to controllably release at least some of the said sample ions contained within the said volume of the storage device

from a storage device exit, the release voltage being of a magnitude such that the potential difference then experienced by the ions across the volume $\boldsymbol{\rho}$ is greater than the said average kinetic energy \overline{E}_k during the trapping period or said part thereof, and further wherein the release voltage is such 5 that the strength of the electric field generated thereby at any first point across the volume p, upon application of the said release voltage, is no more than 50% greater or smaller than the strength of the electric field generated thereby at any other second point across the volume ρ ; (e) receiving those 10 sample ions released from the storage device exit according to the criteria of step (d) through an entrance of an electrostatic trap having a plurality of trapping electrodes, the ions arriving as a convolution of bunched time of flight distributions for each m/z, each distribution having a full width at 15 half maximum (FWHM); and (f) trapping the received sample ions within the electrostatic trap by applying a potential to the electrodes such that the sample ions describe movement having periodic oscillations in at least one direction.

The method of the present invention proposes particular restrictions on the release potential supplied to the storage device which ensures that groups of ions of a given m/z arrive as a tightly focussed 'bunch' at or adjacent the electrostatic trap. The two conditions are, respectively, that ²⁵ the electric field experienced by the ions upon leaving the storage device is relatively uniform and that the potential drop experienced by the ions upon release is larger than the average thermal (kinetic) energy of the ions when trapped in the storage device, and preferably much larger.

The first condition is a consequence of the effective focussing of ions in time of flight from the storage device and into the electrostatic trap. The focal length, L, is given by $L=\alpha V/E$, where V is the final energy of the ions when 35 released from the storage device, E is the electric field strength, and α is a constant. If the electric field is nonuniform, therefore, along the volume ρ , then ions of the same m/z will be focussed at different lengths L and will not arrive in the same tightly focussed bunch. A variation in the 40 electric field strength of no more than about 30%-50% is preferred.

The second criterion ensures that the thermal energy (i.e., the kinetic energy) of the trapped ions is less than and preferably insignificant relative to the gradient of the poten- $_{45}$ tial 'slope' upon which the ions find themselves when the release potential is applied.

In an idealized storage device or ion trap, the ions are all in the same place and leave with exactly the same energy when ejected. This means that they arrive at the same time $_{50}$ at any chosen location downstream of the ion trap. In reality, of course, the ions have a range of kinetic energies and start off from different locations within the trap. Hence the ions of the same m/z arrive at different times downstream of the ion trap. The purpose of TOF focussing is therefore to cause 55 ions further back in the trap to 'catch up' with ions ejected from the front of the trap, by ensuring that the ions nearer the front of the trap move more slowly than those leaving later. The release potential is chosen so that the ions are ejected without being affected significantly by the random pertur- 60 bations of thermal energy spread. Although it is necessary that the potential drop is at least as much as the average kinetic energy, a multiple of two is preferred, of five is more preferable, and at least one order of magnitude is most preferable. 65

This second condition provides the further advantage that the ions will be relatively energetic upon arrival at or adjacent the electrostatic trap. Electrostatic traps have an energy acceptance window, which is centred at a relatively high energy, so that the ions ejected in accordance with the conditions of the present invention are within that energy acceptance window.

Provided that the foregoing conditions are met, the ions will arrive at the electrostatic trap as a convolution of short, energetic packets of similar m/z. Such packets are ideally suited to an electrostatic trap (and particularly the preferred embodiment of an orbitrap) because the FWHM of each ion packet's TOF distribution for a given m/z ratio is then less than the period of oscillation of those sample ions having that m/z when in the electrostatic trap. In other words, the packets are sufficiently coherent for detection to take place.

In preferred embodiments, the ions are pre-cooled, for example in the storage device. This reduces the thermal energy of the ions and also their energy spread, hence reducing the ratio of kinetic energy to release voltage which is the second prerequisite set out above. Pre-cooling may be achieved by collisional cooling, for example. Furthermore, the trapping voltage may be applied so as to force the ions in the storage device towards the exit thereof. This may either be carried out throughout the trapping period or may, in preference, be carried out only immediately prior to ejection.

To obtain bunching of the sample ions at the electrostatic trap, it is preferable to employ an axially segmented linear trap (LT) as the storage device. A differential, preferably DC, voltage is applied between the two or more segments of the LT so as to force the ion cloud within the LT towards the exit thereof. This procedure may be carried out after trapping and cooling of the sample ions received from an ion source in the storage device, which may be achieved by applying voltages to the pole segments so as to create an axial potential well whose base is in the middle of the LT. Alternatively, the bottom of the potential well may be located from the outset at or towards the exit from the storage device. It is particularly preferable that the bottom of the potential well is no more than twice the diameter inscribed by the poles of the storage device away from the storage device exit.

The ions in the storage device are preferably gated out of the exit thereof by applying one or more voltage pulses, to an end electrode of the storage device for example.

The condition for correct focussing or bunching of the sample ions at or adjacent the electrostatic trap is met, in preferred embodiments, by the requirement that the period of oscillation of those ions when injected into the electrostatic trap is shorter, and preferably much shorter, than the time of flight of those sample ions between the storage device exit and arrival at the electrostatic trap entrance.

In alternative embodiments, the further condition that the FWHM time of flight distribution of the ions arriving at the electrostatic trap should be less than the TOF along each detection electrode in the electrostatic trap is imposed as well.

In a particularly preferred embodiment, the storage device is arranged to receive ions along a first direction and to release them from the storage device along a second, orthogonal direction. This permits much higher space charge capacity and better ion beam parameters.

In that case, it is preferable that the storage device should be curved along the first direction. This improves geometric focussing. Optionally, lenses may be added to convert a wide angle beam into a narrow beam. The method may also include boosting the ion energies prior to their arrival at the electrostatic trap entrance.

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In preferred embodiments, the method is employed for the generation, storage and detection of sample ions in MS-only mode. Alternatively, the method may be employed for collision-induced dissociation of the ions to produce daughter sample ion fragments. In either case, it is preferable to 5 select the release voltage so as to focus the ions on the entrance to the orbitrap. In an alternative embodiment, however, the method may be employed to detect fragment ions using surface-induced dissociation. In this case, the method preferably further comprises focussing the sample 10 ions through the electrostatic trap and onto a collision surface. The fragment ions which result are then accelerated back towards the electrostatic trap.

It is particularly preferable that the ions arrive at the electrostatic trap at an angle tangential to a central plane of ¹⁵ the electrostatic trap. This is preferably achieved by lenses between the ion trap and the electrostatic trap. The benefit of this is that no further excitation of the ions is necessary once they enter the electrostatic trap. This in turn reduces the amount of electronics necessary for correct operation of the ²⁰ electrostatic trap, and is to be compared with the arrangement of U.S. Pat. No. 5,886,346 referenced above.

The lenses between the ion trap and electrostatic trap, where present, preferably offer no direct line of sight between the inside of the ion trap and the inside of the electrostatic trap. This arrangement prevents streaming of ions and gas carryover from the (relatively high pressure) ion trap into the (relatively lower pressure) electrostatic trap.

Detection of the sample ions in the electrostatic trap may be achieved in a number of ways. Most preferably, the electrostatic trap is of the orbitrap type and ions are trapped in a hyper-logarithmic field. As bunches of coherent ions of different m/z pass by the outer electrodes of the orbitrap, an image current is induced therein. This current may be amplified and then processed to generate a TOF spectrum, ³⁵ for example by Fourier transform analysis.

The field within the electrostatic trap may preferably be compensated by applying a compensating voltage (which may be time dependent) to a field compensator during detection of the ions. This procedure ensures minimum field perturbation within the volume occupied by the ion trajectories. Additionally or alternatively, during ion injection into the electrostatic trap, the field compensator may be arranged to act as a deflector to improve the trapping efficiency of the electrostatic trip.

As with previous ion traps, and LTs in particular, the ion trap may contain facilities for resonance or mass-selective instability scans to provide for data-dependent excitation, fragmentation or elimination of certain m/z ratios.

The optimum duration of ion trapping in the ion trap may be determined prior to commencement of mass analysis by carrying out a pre-scan. Preferably, a secondary electron multiplier (SEM) or the like is employed. The SEM may be located radially of the ion trap and in that case massselective instability or a resonance excitation scan in the ion trap may be used. Most preferably, however, an axial SEM is employed downstream of the electrostatic trap and on an ion beam axis. In this case, ions are preferably injected into the electrostatic trap just as they would be for subsequent 60 mass analysis.

According to a second aspect of the present invention, there is provided a mass spectrometer comprising: (a) an ion source arranged to supply a plurality of sample ions to be analysed, each of which has a mass-to-charge ratio m/z; (b) 65 an ion storage device comprising a plurality of storage device poles and having a storage device entrance end 6

through which the said sample ions are received and a storage device exit end through which the said sample ions may exit; (c) a voltage source arranged to supply a trapping voltage to the storage device poles so as to contain at least a proportion of the sample ions received through the storage device entrance end of the storage device within a volume p of the storage device in a trapping mode during at least a part of a trapping period, the thus trapped ions each having a kinetic energy E_k such that there is an average kinetic energy $\overline{\mathbf{E}}_k$ of the ions in the volume ρ during the trapping period or the said part thereof, and to supply a release voltage to the storage device in an ion ejection mode so as to controllably release at least some of the said sample ions contained within the said volume ρ of the storage device through the storage device exit end, the release voltage being of a magnitude such that the potential difference then experienced by the ions across the volume ρ is greater than the said average kinetic energy \overline{E}_k during the trapping period or said part thereof, and further wherein the release voltage is such that the electric field generated thereby at any first point across the volume ρ , upon application of the said release voltage, is no more than 50% greater or smaller than the electric field generated thereby at any other second point across the volume ρ ; and (d) an electrostatic trap having an electrostatic trap entrance arranged to receive those ions released through the storage device exit end and meeting the criteria imposed by the applied trapping and release potentials, as a convolution of bunched time of flight distributions for each m/z, each distribution having a full width at half maximum (FWHM); the electrostatic trap further comprising a plurality of electrodes arranged to trap ions received through the electrostatic trap entrance therebetween so that the said trapped ions describe movement having periodic oscillations in at least one direction.

As with the method of the first aspect, a segmented multipolar ion trap is preferable. In that case, to maximise focussing or bunching of ion packets at the entrance to the electrostatic trap, the length of the segment of the pole pieces proximal the ion trap is preferably shorter than twice the inscribed diameter between the segments of the ion trap (measured radially), and most preferably shorter than the inscribed diameter. Also, the distance between the ion trap exit and the centre of the segment closest thereto is preferably greater than or equal to the said inscribed diameter.

Other features and advantages of the invention will become apparent with reference to the appended claims and to the following specific description.

BRIEF DESCRIPTION OF THE DRAWINGS

The present invention may be put into practice in a number of ways, and some specific embodiments will now be described by way of example only and with reference to the accompanying drawings in which:

FIG. 1 is a schematic side view of a mass spectrometer embodying the present invention and including an ion trap and electrostatic trap;

FIG. 2 shows schematically and in perspective a part of the ion trap of FIG. 1;

FIG. **3** shows a front view of the electrostatic trap of FIG. **1**;

FIG. 4a shows, schematically, the potential distribution in the ion trap of FIGS. 1, 2 and 3 when ions are trapped therein;

FIG. 4b shows, schematically, the potential distribution in the ion trap of FIGS. 1, 2 and 3 at the point when trapped ions are ejected from the ion trap;

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FIG. 5 shows a front view of an alternative electrostatic trap for use in the mass spectrometer of FIG. 1;

FIG. 6 shows an alternative configuration of the mass spectrometer of FIG. 1, again in schematic side view and including a curved ion trap along with an electrostatic trap; 5

FIG. 7 shows a sectional view through the curved ion trap of FIG. 6: and

FIG. 8 shows a side view of an electrostatic trap for use with the mass spectrometer of FIG. 1 OR FIG. 6, in 10 combination with a collision surface downstream of the electrostatic trap to allow operation of the mass spectrometer in surface-induced dissociation mode.

Referring first to FIG. 1, a mass spectrometer 10 is shown. The mass spectrometer comprises a continuous or pulsed ion source 12, such as an electron impact source, an electrospray source (with or without a collisional RF multipole), a matrix-assisted laser desorption and ionisation (MALDI) source, again with or without a collisional RF multipole, and so forth. In FIG. 1, an electrospray ion source 12 is shown.

The nebulized ions from the ion source 12 enter an ion source block 16 having an entrance cone 14 and an exit cone 18. As is described, for example, in WO 98/49710, the exit cone 18 has an entrance at 90° to the ion flow in the block 16 so that it acts as a skimmer to prevent streaming of ions $_{25}$ into the subsequent mass analysis components.

A first component downstream of the exit cone 18 is an ion cooler 20 which reduces the energy of the sample ions from the ion source 12. Cooled ions exit the ion cooler 20 through an aperture 22 and arrive at a quadrupole mass filter 30 24 which is supplied with a D.C. voltage upon which is superimposed an arbitrary r.f. signal. This mass filter extracts only those ions within a window of m/z of interest and the chosen ions are then released to a linear ion trap 30. The ion trap **30** is segmented, in the embodiment of FIG. 1, 35 into an entrance segment 40 and an exit segment 50. Although only two segments are shown in FIG. 1, it will be understood that three or more segments could instead be employed. As better seen in FIG. 2, the segments 40, 50 are each formed from four rods which are radially spaced so as $_{40}$ to form a trapping volume 60 between them.

To trap ions within the trapping volume 60, a voltage source (not shown) applies an RF voltage to each of the segments 40, 50. The application of an RF field generates a potential well in the axial direction. Collisions between ions 45 entering the linear trap 30 rapidly cause these ions to sink towards the bottom of the potential well.

The ends of the linear trap 30 are bounded by exit and entrance electrodes 70, 80 respectively. These electrodes are supplied with a DC voltage V_D and V_a respectively. As will 50 be familiar to those skilled in the art, the linear trap 30 may also contain facilities for resonance or mass-selective instability scans, to provide data-dependent excitation, fragmentation or elimination of selected mass-to-charge ratios.

In preference, the length of the exit segments 50 is not in 55 excess of the inscribed diameter D between the rods (FIG. 2). Also, the distance x between the exit electrode 70 and the axial centre of the exit segment 50 is preferably comparable to, or greater than, the inscribed diameter D (again, see FIG. 2).

The linear trap 30 may have a pressure gradient therein. In this way, the conditions in one part of the trap 30 are optimised for the best dissipation of energy through ion collisions, whilst near the exit electrode 70, the conditions may be optimised for the best trapping, lowest fragmenta- 65 tion and so forth. The pressure gradient may, for example, be created through the introduction of additional gas inlets.

Downstream of the exit electrode is a deflection lens arrangement 90 including deflectors 100, 110. The deflection lens arrangement is arranged to deflect the ions exiting the linear trap 30 in such a way that there is no direct line of sight connecting the interior of the linear trap 30 with the interior of an electrostatic orbitrap 130 downstream of the deflection lens arrangement 90. This prevents streaming of energetic ions from the relatively high pressure linear trap 30 into the relatively low pressure orbitrap 130. The deflection lens arrangement 90 also acts as a differential pumping aperture.

Downstream of the deflection lens arrangement 90 is a conductivity restrictor 120. This sustains a pressure differential between the orbitrap 130 and the lens arrangement 90.

Ions exiting the deflection lens arrangement 90 through the conductivity restrictor 120 arrive at the orbitrap 130. The orbitrap 130 has a central electrode 140 as may better be seen with reference now to FIG. 3. The central electrode 140 is connected to a high voltage amplifier 150.

The orbitrap 130 also contains an outer electrode split into two outer electrode parts 160, 170. Each of the two outer electrode parts 160, 170 is connected to a differential amplifier 180. Preferably, this differential amplifier is maintained at virtual ground.

Referring once more to FIG. 1, downstream of the orbitrap 130 is a secondary electron multiplier 190, on the optical axis of the ion beam. Although not shown in FIG. 1, the secondary electron multiplier (SEM) 190 may also be located on the side of the linear trap 130.

The system, and particularly the voltages applied to the various parts of the system, is controlled by a data acquisition system. This data acquisition system is in itself known and does not form a part of the present invention. Accordingly, it is not shown in the Figures and will not be described further. The data acquisition system may also carry out signal processing as described below. Likewise, a vacuum envelope is also provided, to allow differential pumping of the system. Again, this is not shown in the Figures, although the typical pressures are indicated in FIG. 1.

In operation, ions from the ion source 12 enter the segmented linear trap 30 and are reflected by an elevated potential V_D on the exit electrode 70 thereof. AC voltages at RF frequencies are applied to the segments of the trap to provide a quasi-potential well in the radial direction whilst DC voltages V_a , V_b and V_c provide a potential well along the axis of the linear trap 30. The pressure inside the linear trap 30 is chosen in such a way that ions lose sufficient kinetic energy during their first pass through the trap that they accumulate near the bottom of the axial potential well. Before ions are removed from the linear trap 30, the DC voltages V_a , V_b , V_d and V_D may be varied in such a way that the centre of ion cloud within the linear trap 30 is shifted into the end section of the linear trap, that is, into the volume defined between the rods in the exit segment 50 adjacent to the exit electrode 70. As an alternative, the bottom of the axial potential well may instead be located in this exit segment 50 from the start of ion storage in the linear trap 30.

FIG. 4a shows, not to scale, the potential electrode 80 and the exit electrode 70 when ions are trapped therein. It will be seen that the ions sit in a potential well defined by the difference in potentials between the exit electrode 70, the exit segment $\overline{50}$ of the trap 30 and the entrance segment 40thereof.

At the end of storage, the data acquisition system starts to ramp the voltage applied to the central electrode 140 in the

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orbitrap 130 and, simultaneously, applies a voltage pulse to the exit electrode 70 of the linear trap 30. In presently preferred embodiments, a single pulse is applied to empty the linear trap. However, multiple pulses may be employed instead. In any event, the delay between successive pulses is chosen in such a way that all of the mass range of interest arrives into the orbitrap 130 during the correct phase of the voltage applied to the central electrode 140 thereof as it is ramped. Although the ramping of the voltages on the central electrode of the orbitrap 130 and the exit electrode 70 of the linear trap are timed to each other, they do not however need to be synchronous. Thus, the voltage applied to the central electrode 140 of the orbitrap 130 may start to ramp before the pulse is applied to the exit electrode 70 on the linear trap, and may continue to ramp for a period (e.g. tens of microseconds) after the linear trap has been emptied.

FIG. 4*b* shows the potential at the different points along the ion trap **30** between the entrance **80** and the exit electrode **70** again not to scale, when such a pulse is applied²⁰ to the exit electrode **70**. Because the pulse is negative (in the convention adapted to illustrate the specific embodiment described, the ions previously trapped in the potential well instantaneously find themselves on a "slope" which accelerates them away from the ion trap **30**. As will be explained below, the ejection technique causes the ions leaving the ion trap **30** to be time-of-flight focussed onto the entrance of the electrostatic trap **130**.

It is important, for correct trapping of ions in the orbitrap ³⁰ **130**, that they arrive at the entrance to the orbitrap when the voltage on the central electrode **140** thereof is between approximately $(D_1/D_2)^{1/2}$ V, and V, where V is the final, static voltage on the central electrode **140**, D_1 is the outer diameter of the central electrode **140**, and D_2 is the inner diameter of the outer electrode formed from the outer electrode parts **160** and **170**. The ion energy at the entrance to the orbitrap **130** also needs to lie within a certain range.

Whilst the voltage applied to the central electrode 140 of 40 the orbitrap 130 is ramped, ions are directed and focussed by the linear trap 30 and the deflection lens arrangement 90 to the entrance of the orbitrap 130. The ions enter the field within the orbitrap 130 tangentially to the outer electrodes formed from the outer electrode parts 160, 170 and are prevented from hitting this electrode again by a monotonically-increasing electric field, which squeezes the ions closer to the centre of the trap. Tangential injection into the orbitrap 130 is achieved by displacing the trap relative to $_{50}$ the centre of the beam of ions arriving from the ion trap 30. By way of example only, the orbitrap 130 may be positioned so that ions enter it at a radius of 17.4 mm with z=10 mm, the highest internal radius of the electrodes being 20 mm in this specific example. 55

The rise time of the electric field depends on the mass range to be trapped, ion parameters, and the orbitrap **130**, but is usually between 20 and 200 microseconds. Squeezing stops when there is no more threat of losing ions onto the electrodes.

The orbitrap 130 is shaped so as to generate a hyperlogarithmic field between the central electrode 140 and the outer electrode formed from the outer electrode parts 160, 170. The potential distribution of this hyper-logarithmic 65 field may be described in cylindrical coordinates (r, z) by the following equation:

$$V(r, z) = \frac{k}{2} \left[z^2 - \frac{r^2}{2} \right] + \frac{k}{2} (R_m)^2 \ln \left[\frac{r}{R_m} \right] + C$$

where z=0 is the plane of symmetry of the field, C, k, $R_m(>0)$ are constants, and k>0 for positive ions. Such a field creates a potential well along the z axis direction which causes ion trapping in that potential well provided that the incident energy is not too great for the ion to escape. As the voltage applied to the centre of electrode **140** increases, the field intensity increases and therefore the force on the ions towards the longitudinal axis increases, thus decreasing the radius of spiral of the ions as may be seen from FIGS. 1 and **2**. Thus, the ions are forced to rotate in spirals of smaller radius as the sides of the potential well increase in gradient.

There are three characteristic frequencies of oscillation within the hyper-logarithmic field. The first is the harmonic motion of the ions in the axial direction where they oscillate in the potential well with a frequency independent of energy in this direction. The second characteristic frequency is oscillation in the radial direction since not all of the trajectories will be perfectly circular. The third frequency characteristic of the trapped ions is the frequency of angular rotation.

Further details of the preferred electrode arrangement of the orbitrap **130** may be found in U.S. Pat. No. 5,886,346, referenced above, the contents of which are incorporated by reference in their entirety. It will, however, be noted that, in the present case, the ions enter the field tangentially and do not require a separate injection of radial force which in turn reduces the amount of electronic control of the orbitrap **130** that is necessary.

The ion packets arriving at the entrance to the orbitrap 130 are bunched together due to the time of flight focussing created by the ion ejection technique from the linear trap 130. The ion packets are sufficiently coherent that coherent axial oscillation within the orbitrap 130 takes place without addition excitation. The required degree of coherency depends on the type of detection.

If all ions of the same mass-to-charge ratio were to have the same initial kinetic energy, begin flight at the same time, and from the same position within the trap, then they would all leave the trap together and travel together to arrive at exactly the same moment at any point downstream of the trap. This idealized situation cannot of course be realized in practice, primarily due to three factors which 'smear' the initial peak from a delta function. Firstly, the starting position of different ions of the same mass-to-charge ratio will be different, secondly, the time at which flight begins, and thirdly the initial kinetic energies of the different ions of the same mass-to-charge ratio will be different.

The present invention addresses the non-ideal nature of the ions in the trap by firstly minimizing the spread of initial kinetic energies and by 'bunching' the ions together at one end of the trap (so that they tend to leave from roughly the same point in the trap) and secondly by focussing the ions during flight so that any temporal, spatial or energetic spread which still remains is reduced. FIG. **4***b* shows why this should be so.

When the exit electrode 70 receives the negative going pulse to eject the ions, those ions of the same m/z, which at the moment of pulsing are closest to the exit electrode 70, experience a smaller voltage drop than those further away from the exit electrode 70. As the ions of similar m/z which are closest to the exit aperture have a smaller distance to travel to that exit aperture, they pass through it earlier than the ions 'behind' them but at a lower velocity. In other

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words, ions further away from the exit aperture 70 on application of the negative pulse take longer to be emptied from the tap but leave it at a higher velocity. In this manner, the ions bunch up downstream of the ion trap 30. By carefully selecting the ejection parameters, the ions in the 5 ion trap 30 may be focussed onto the entrance of the orbitrap 130.

In the preferred embodiment shown in FIGS. 1 and 3, a mass spectrum is generated using image current detection, which technique is again described in U.S. Pat. No. 5,886, 10 346. An interference pattern of frequencies of different mass-to-charge ratios produces an image current on the outer electrode parts 160, 170. This current is amplified by the differential amplifier 180 and then processed by the data acquisition system by application of a Fourier transform. For this type of ion detection, coherency of the ion packets is achieved as soon as the duration dT (m/z) of a given ion packet of a specific m/z is smaller than the period of axial oscillations within the orbitrap 130. This period of axial oscillation may in turn be significantly less than the time of 20 flight between the linear trap 30 and the entrance to the orbitrap 130. To achieve this level of coherence, time of flight focussing or bunching of the ions as they enter the orbitrap 130 is necessary. It is important to distinguish this bunching from first or second order focussing typical in time 25 of flight (TOF) mass analysers. The time of flight between the linear trap 30 and the orbitrap 130 is normally significantly greater than the period of axial oscillations within the orbit trap 130 because of the differential pumping between the linear trap 30 and the orbitrap 130 which necessarily 30 requires a significant spatial separation between the linear trap and the orbitrap 130.

The electric field strength and the ion trap dimensions are chosen in such a way that the ions catch up, that is, focus in time of flight, just at the entrance to the EST 130. For 35 example, for a uniform field, the focal point is located 3.d away from the starting point where d is the length of this uniform field as measured from the starting point to the field boundary. Non-uniform fields have focussing properties but the mathematical relations are very complex. The random 40 energy distributions of the ions trapped in the ion trap 30 adversely affect the quality of the TOF focussing; for example, the pulse width becomes comparable with the total time of flight as soon as the energy spread of the ions becomes comparable with the acceleration voltage. In order 45 to address these problems, therefor, the bunching of the ions is achieved, in practice, through a combination of the following requirements.

Firstly, the relative variation of the electric field strength along the ion cloud within the ion trap 30, or at least the 50 described in the above-referenced U.S. Pat. No. 5,886,346 portion to be injected into the orbitrap 130, should be smaller than unity, and preferably much smaller than unity (such as 10% or even less). Such uniformity of the field along the beam may be achieved by ion squeezing prior to pulsing of the linear trap **30** which in turn is achieved by moving the base of the potential well in the linear trap 30 towards the exit electrode 70 as described above. Collisional cooling of ions within the linear trap also assists in this squeezing. Secondly, the voltage drop across the ion cloud in the linear trap should at least exceed the average kinetic 60 energy during storage, and preferably exceed it by an order of magnitude or more. Such considerable field strength reduces the time of flight spread caused by the initial energies.

The optimum duration of trapping of ions within the 65 linear trap 30 may be determined using a pre-scan after a short storage duration. The secondary electron multiplier

190 allows detection. Where the SEM 190 is located radially of the linear trap 30, mass-selective instability or a resonance excitation scan in the linear trap 30 may be used. It is preferable, however, to use an axial SEM 190, after the orbitrap 130, so that ions are injected in the same way as for the analysis in the orbitrap 130. It will be understood by those skilled in the art that any other known way of ion detection could be used instead of SEM 190, such as a collector with a charge-sensitive amplifier, a photomultiplier, semiconductor detectors and so forth.

During detection of ions in the orbitrap 130, an appropriate voltage, which may be time-dependent, may be supplied to a field compensator 200 adjacent the entrance to the orbitrap 130. This field compensator ensures minimum field perturbation within the volume occupied by the ion trajectories. During ion injection, this field compensator 200 acts also as a deflector to improve trapping efficiency within the orbitrap 130.

Although a preferred embodiment has been described, it will be understood that various modifications are contemplated. For example, although a linear ion trap 30 has been described for storage of the sample ions from the ion source 12, it is to be understood that a quadrupole ion trap could equally be employed for ion storage, cooling and so forth. Quadrupole ion traps are known as such and one example is shown in EP-A-0,113,207.

Furthermore, a second form of orbitrap 130, seen in front view in FIG. 5, may be employed instead of the one shown in FIG. 3. Here, instead of split outer electrode parts, the outer electrode is not split. Instead, the central electrode 140' is axially segmented with a centre part 220 connected to a pre-amplifier 210. In this arrangement, pulses of image current from passing ion packets of different m/z are detected by this central, disc-shaped electrode part 220, amplified by the pre-amplifier 210, and then processed to yield a time of flight spectrum. A deconvolution method such as is described by M. May et al, in Analytical Chemistry (1992) vol. 64, pages 1601-1605 could be used, or any other signal reconstruction method. Two or more detection electrodes could also be used. The pre-amplifier could also be a differential pre-amplifier with a second input connected to another detection electrode or simply open-ended to improve common mode rejection. For this type of detection, the best results are achieved when the duration dT (m/z) of each ion packet of a particular m/z is not only smaller than the period of axial oscillations in the orbitrap 130, but also does not exceed the time of flight along each of the detection electrodes.

Furthermore, a mass selective instability scan, such as is could also be used. In this case, ion injection could be performed also along the central plane of the trap.

FIG. 6 shows another mass spectrometer 10' which embodies the present invention and currently represents the preferred implementation. Features common to FIGS. 1 and 6 are labelled with like reference numerals.

As with the arrangement of FIG. 1, the mass spectrometer 10' comprises an electrospray ion source 12 which provides nebulized ions into an ion source block 16 through an entrance cone 14. The ions exit the block 16 via an exit cone 18 and pass into an ion cooler 20 at around 10^{-2} mbar. The ions then arrive in a quadrupole mass filter 24 via an aperture 22, and a range of m/z of the incident ions is selected as described previously.

Ions exiting the mass filter 24 enter an ion trap 300 in a first direction generally parallel with a 'y' axis (see FIG. 6). The ions are however ejected from the ion trap 300 in a

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second direction generally orthogonal to their entrance direction, that is, in an 'x' direction as indicated in FIG. 6. As previously, the ions are focussed in time of flight downstream of the linear trap. Orthogonal ejection of trapped ions additionally allows a much higher space charge capacity than is provided by the arrangement of FIG. 1, and also provides better ion beam parameters.

Although, to achieve orthogonal ejection, a quadrupole assembly such as is shown in FIG. 1 could be used, this requires the RF voltages applied thereto to be switched off instantaneously. This requires very complex associated electronics, however, so, in preference, the ion trap **300** is elongate in the 'y' direction but is curved in the x-y plane (such that the longitudinal axis is likewise curved in the x-y plane). Curvature of the ion trap **300** improves geometrical focussing. By way of distinguishing from the truly linear trap **130** of FIG. **1**, the ion trap **300** will henceforth be termed a 'curved trap'.

The curved trap **300** includes lenses **310** extending from the exit of the trap which together convert the wide angle incident beam into a narrow beam.

The narrow ion beam exiting the curved trap **300** passes through a beam deflector **320** and into an electrostatic trap **130**. The beam deflector may take one of a variety of forms. A plate in front of the focal point of the focussing lenses could be used to block both ions and gas along the line of 25 sight towards the entrance to the electrostatic trap **130** (for example, around $\pm 5^{\circ}$ of arc). Gas arriving from larger angles will not be along the line of sight whereas corresponding ions can be diverted into the electrostatic trap entrance by lenses. The problem with this approach is that it requires the 30 blocking plate to be located in a field-free region which can be difficult to arrange. As an alternative, a toroidal deflector can be employed to achieve the required beam deflection although this introduces extra complexity.

The preferred beam deflector **320** is shown in FIG. **6** and 35 contains a right-angled bend which prevents gas carryover along the line of sight between the curved trap **300** and the electrostatic trap **130**. An 'S'-shaped beam deflector could of course be employed instead (as is shown in FIG. **1**). If the electrostatic trap is arranged with its entrance parallel to the 40 direction of exit of ions from the curved trap **300**. As with the arrangement of FIG. **1**, the ions are usually focussed in time of flight onto the electrostatic trap entrance which then detects in the manner described previously. However, a different focal point can be chosen, as for example when 45 operating in surface-induced dissociation mode (see FIG. **8** below).

Referring now to FIG. 7, a sectional view of the curved trap 300 of FIG. 6 is shown. The curved trap 300 is preferably RF only and comprises an outer electrode plate 50 330, at a D.C. voltage preferably near ground, along with an inner electrode plate 340 at the same D.C. voltage. Sandwiched between the inner and outer electrode plates 330, 340 are upper and lower centre plate pairs 360*a*, 360*b*, to which is applied an RF voltage having a phase indicated on 55 the drawings as RF–. The upper and lower centre plate pairs 360*a*, 360*b* in turn sandwich an axis plate pair 350 to which is applied an RF voltage (labelled RF+ in FIG. 7) which is in antiphase to the voltage applied to the upper and lower centre plate pairs 360*a*, 360*b*. 60

The geometry of the curved trap 300 is chosen in such a way that the minimum of the effective potential (that is, the minimum of the RF field) is located exactly in the middle of the trap. This is the axis of symmetry in the plane XZ and is labelled point Q in FIG. 7.

The electrodes **330**, **340**, **350**, **360***a* and **360***b* are curved in the XY plane. A DC offset, which is the same as that 14

applied to the inner and outer electrode plates 330, 340, is applied to the upper, lower and axis plates. This causes all masses to be stored in the curved trap 300 and cooled in collisions with gas at 0.1-1 mtorr. At the end of the storage, a positive pulse is applied to the outer electrode plate 330. and a negative pulse of the same amplitude is applied to the inner electrode plate 340 (for positive ions). Ions are extracted by the resulting electric field through a trap exit 370. The RF voltages do not need to be removed as they have little effect on the beam parameters due to their symmetry. In addition, the RF field strength near the X axis is relatively weak. However, it is preferable to time the pulses applied to the inner and outer electrode plates 330, 340 so that they are applied in synchronism with the phase of the RF voltages applied to the upper, lower and axis plates 350, 360a, 360b.

After sufficient ions have been stored in the trap, they are ejected towards the centre of curvature of the curved trap **300** (see below), and also focussed, both geometrically and in time-of-flight, into a narrow beam that is then introduced into the electrostatic trap **130**.

Still referring to FIG. 7, a liner 380 may be provided between the trap exit 370 and the lenses 310. The liner 380 is located in a substantially field-free region of the curved trap. Another pulse, equal to the required ion energy on the entrance to the electrostatic trap 130, may be applied to the liner 380 at the same time as the pulses are applied to the inner and outer electrode plates 330, 340. The pulse is applied to the liner so as to create an "energy lift", that is, to produce a high ion energy in conditions where both the DC offset applied to the curved trap 300 and the potential applied to the outer electrode parts 160, 170 of the orbitrap 130 (FIG. 3) are maintained near ground voltage. If the curved trap 300 floats at the acceleration voltage then no energy lift will be required. Nevertheless, it is important that any ion source should have the capability to float as well.

The length of the liner **380** is defined by the required relative mass range: the ratio of the maximum to minimum masses is given by $m_{max}/m_{min}=(L_1/L_2)^2$, where L_1 is the effective distance from the axis to the exit from liner and L_2 is the effective distance to the entrance to the liner. The duration of the pulse applied to the liner **380** is determined by calculating the time-of-flight, to the liner exit, for the lightest mass of interest, so that this mass emerges from the liner **380** at its full energy whilst the voltage on the liner **380** is already back to its normal value (near ground). The duration of the pulses applied to the inner and outer electrode plates **330**, **340** should be longer than this.

It is to be understood that the liner **380** in the curved trap **300** of FIGS. **6** and **7** is equally suitable for the linear trap **30** of FIG. **1**.

The foregoing description of some preferred embodiments has also explained the principles involved with reference to sample ions that are derived directly from an atmospheric pressure ionization source or the like, and are simply stored as such in the ion trap. However, structural analysis of sample ions may also be carried out using TOF focussing and any of the ms/ms modes available. Three modes in particular are contemplated.

In collision-induced dissociation (CID), precursor ions may be selected either by the quadrupole mass filter **24** (FIG. **1**) or inside the ion trap **30**. Ejection of unwanted ions in each of these cases may be performed, for example, by resonant excitation between the opposite rods of each device or by a mass selective instability scan (see, for example, the above-referenced U.S. Pat. No. 5,886,346). This may be achieved by DC biassing one set of rods relative to the other, for example. Fragmentation may be caused by collisions with collision gas at an elevated pressure in a dedicated RF-only multipole or, preferably, in the linear ion trap 30. The resulting fragments are stored and collisionally cooled in the ion trap 30 and then injected into the orbitrap 130 in 5 the same way as described previously. CID in collisional multipoles is a technique which is well known as such to those skilled in the art and the technique of selecting only the required m/z is likewise a known part of tandem mass spectrometry. Further description of these techniques may 10 be found in "Protein Sequencing and Identification Using Tandem Mass Spectrometry", by M. Kinter, N. E. Sherman, John Wiley and Sons, 2000, and in "Mass Spectrometry/ Mass Spectrometry: Techniques and Applications of Tandem Mass Spectrometry", by K. L. Busch, G. L. Glish, and 15 S. A. McLuckey, John Wiley and Sons, 1989.

In surface-induced dissociation (SID), precursor ion selection is typically carried out in the same way as in CID. Precursor ions are stored in the linear trap 30 and then pulsed towards the orbitrap 130. However, in this case the TOF 20 focus is shifted behind the plane of the entrance of the orbitrap 130, to a separate plane where a collision surface is located. This is shown schematically in FIG. 8, which illustrates in side view the orbitrap 130 along with the electrode 140, the compensator 200 and the secondary 25 electron multiplier 190. The collision surface 250 is located downstream of the secondary electron multiplier 190 and may be formed from a metal or may instead be metal- or polymer-coated. A fluorocarbon or hydrocarbon selfassembled monolayer plate such as is disclosed in Science, 30 Vol. 275, pages 1447-1450, by S. A. Miller, H. Luo, S. J. Pachuta and R. G. Cooks, (1997) could for example be used.

In this case, precursor ions pass tangentially through the orbitrap 130, which has electric fields that are low enough to prevent ion losses, and out past the compensator 200 which 35 in this part of the process is switched off to allow passage of the precursor ions. These precursor ions then decelerate in front of the collision surface 250 in a deceleration gap created by a grid 255 and collide with the collision surface **250** at a collision energy determined by a voltage difference 40 between the collision surface and the offset applied to the exit segment 50 of the linear trap. Collision results in the formation of fragment ions at low energies (several electronvolts) which are accelerated by the same electric field towards the orbitrap 130. Due to the TOF focussing of 45 these precursor ions and the instantaneous nature of SID, fragment ions separate (at least partially) on their way to the orbitrap 130 according to their mass/charge ratio and ions of each mass/charge ratio enter the orbitrap 130 as a short packet. The compensator 200 is switched on during this part 50 of the process so that the fragment ions are then captured by the orbitrap. This allows ions to be captured in the same way as in the MS-only mode described in connection with FIGS. 1 to 5 above. If low resolution of parent selection is sufficient for a given application, then an ion gate (not shown) may be 55 installed between the orbitrap 130 and the collision surface 250 to provide an alternative way of selecting precursor ions. It is in fact possible to use the deflection electrode 200 of the orbitrap 130 as an ion gate.

Finally, metastable dissociation (MSD) mode may be 60 employed with the arrangement and principles outlined previously. Precursor ions may either be selected as described above in connection with the CID and SID modes, or may instead be injected into the orbitrap **130** without prior selection. The only difference from the MS-only mode 65 described in connection with FIGS. **1** to **5** is the activation of ions in the ion trap **30**. Activation may be achieved by

pulsed ion extraction in the presence of gas at an elevated pressure (either static or pulsed), wherein the increase of ion internal energy is controlled by the gas pressure, by pulsed electromagnetic radiation (e.g. infrared radiation which can be used to excite ions inside the ion trap **30**, or by resonant or broadband dipolar excitation using at least two pairs of quadrupole rod electrodes in the ion trap **30**. In that case, the increase of internal energy is controlled by the amplitude of the excitation signal and the gas pressure.

Although pulsed ion extraction with a high pressure gas is preferable due to its simplicity, each of the foregoing results in the excitation ("heating") of ions and the consequent formation of metastable ions with a controllable decay constant. The magnitude of the decay constant can be controlled by variation of the intensity of excitation. Before fragmentation becomes noticeable, ions may be injected into the orbitrap **130** and precursor ion selection may even be achieved. On the other hand, excessively long decay times lead to a decrease in the speed of analysis. Therefore, optimum decay times range from several milliseconds to tens of milliseconds.

Precursor ion selection is achieved by applying a radio frequency voltage in resonance with the axial oscillation of precursor ions at a correct phase. A waveform generator (not shown), under the control of the data processing system referred to in connection with FIG. 1, supplies this RF voltage either to the central electrode 140 (parametric resonance de-excitation at doubled ion frequency, set out in Analytical Chemistry Volume 72, No. 6, p. 1156-1162, by Makarov, and in the above-referenced U.S. Pat. No. 5,886, 346), or between the two outer electrode parts 160, 170 (resonance de-excitation at ion frequency) of the orbitrap 130. Application of an RF voltage decreases the amplitude of axial oscillation of ions so that only precursor ions are brought into the plane of symmetry of the orbitrap 130. Precursor ions are left in this state long enough to allow metastable decay to occur. The remaining precursor ions are then excited, together with their fragment ions, by a broadband excitation. Typically, a radio frequency voltage is applied to the two outer electrode parts 160, 170 by the waveform generator. Coherent oscillations of ions of each mass/charge ratio are detected by detecting an image detection current via the differential amplifier 180 in the same way as described for MS-only mode. Metastable decay of ions other than precursor ions also results in the formation of fragment ions. However, these are uniformly spaced along the orbitrap 130 and thus do not move coherently. No image current is produced for detection in that case. Alternatively, unwanted precursor or fragment ions may be removed by an additional broadband excitation.

What is claimed is:

1. A method of injection of sample ions into an electrostatic trap, comprising the steps of:

- (a) generating a plurality of sample ions to be analysed, each of which has a mass-to-charge ratio m/z;
- (b) receiving the sample ions through a storage device entrance in an ion storage device having a plurality of storage device poles;
- (c) supplying a trapping voltage to the storage device so as to trap at least a proportion of the received sample ions within a volume ρ in the storage device, during at least a part of a trapping period, the thus trapped ions each having a kinetic energy E_k such that there is an average kinetic energy \overline{E}_k of the ions in the volume ρ during the trapping period or the said part thereof;
- (d) supplying a release voltage to the storage device so as to controllably release at least some of the said sample

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ions contained within the said volume of the storage device from a storage device exit, the release voltage being of a magnitude such that the potential difference then experienced by the ions across the volume ρ is greater than the said average kinetic energy \overline{E}_k during the trapping period or said part thereof, and further wherein the release voltage is such that the strength of the electric field generated thereby at any first point across the volume ρ , upon application of the said release voltage, is no more than 50% greater or smaller than the strength of the electric field generated thereby at any other second point across the volume p;

- (e) receiving those sample ions released from the storage device exit according to the criteria of step (d) through an entrance of an electrostatic trap having a plurality of trapping electrodes, the ions arriving as a convolution of bunched time of flight distributions for each m/z, each distribution having a full width at half maximum (FWHM); and
- (f) trapping the received sample ions within the electrothat the sample ions describe movement having periodic oscillations in at least one direction.

2. The method of claim 1, in which the potential applied to the electrodes of the electrostatic trap is ramped towards the trapping voltage at the same time as the release voltage 25 is applied to the storage device to cause the said controllable release of the sample ions therefrom.

3. The method of claim 2, in which the potential applied to the electrodes of the electrostatic trap commences ramping towards the trapping voltage prior to application of the 30 said release voltage, and continues to be applied until after the release voltage stops being applied.

4. The method of claim 1, in which the trapping voltage includes an AC component to trap the ions in a radial direction of the storage device.

5. The method of claim 4, in which the trapping voltage further comprises a DC component, the trapping voltage being selected so as to create a potential well in the storage device defining the said volume p, the base of which potential well is adjacent to the storage device exit during at 40 least a part of that trapping period.

6. The method of claim 1, in which the release voltage includes at least one DC pulse.

7. The method of claim 1, in which each of the storage device poles is axially segmented into at least two separate 45 pole elements, the step of supplying a trapping voltage to the storage device further comprising supplying a differential DC voltage between at least two separate pole elements so as to cause the sample ions to be trapped in a potential well having a base adjacent to the storage device exit.

8. The method of claim 1, in which the said period of oscillation of those sample ions in the electrostatic trap is shorter than the said FWHM of the time of flight of those sample ions between ejection from the storage device exit and arrival at the electrostatic trap entrance.

9. The method of claim 1, in which, during the step (d), the storage device is arranged to release ions stored therein along an axis generally parallel with a major axis of the storage device when said release voltage is applied.

10. The method of claim 9, in which the storage device is 60 a linear trap whose poles together define an inscribed diameter, and in which the trapping potential is applied so that the base of the potential well is located not more than a distance equal to twice the inscribed diameter from the storage device exit.

11. The method of claim 8, in which the release voltage is a differential voltage applied across the axial segment

adjacent the storage device exit, and wherein the magnitude of the release voltage is greater than the magnitude of the trapping voltage.

 $\hat{12}$. The method of claim 1, in which, during the step (d), the storage device is arranged to release ions stored therein along an axis generally orthogonal with a major axis of the storage device when said release voltage is applied.

13. The method of claim 12, further comprising trapping the ions in a storage device which is curved along its major axis

14. The method of claim 1, further comprising applying a voltage to an energy lifter downstream of the storage device, so as to increase the ion energies as they pass through the said energy lifter.

15. The method of claim 1, further comprising cooling the sample ions to reduce their kinetic energies E_{μ} .

16. The method of claim 15, in which the cooling is carried out in an ion cooler prior to receipt through the said storage device entrance.

17. The method of claim 15 in which the cooling is carried static trap by applying a potential to the electrodes such 20 out within the storage device by forcing sample ions to collide with a collision gas.

> 18. The method of claim 1, in which the release voltage is of a magnitude such that the potential difference then experienced by the ions across the volume ρ is at least an order of magnitude greater than the said average kinetic energy E_k .

> 19. The method of claim 1, in which the release potential is applied so as to release the ions trapped within the volume ρ such that they focus in time of flight upon the entrance to the electrostatic trap.

> 20. The method of claim 19, further comprising directing the ions towards the entrance of the electrostatic trap so that they arrive there at an angle which is tangential to a central plane of the electrostatic trap.

> 21. The method of claim $\hat{1}$, in which the release potential is applied so as to release the ions trapped within the volume ρ such that they focus in time of flight upon a collision surface located downstream of the electrostatic trap, the method further comprising accelerating fragment ions, created by the impact of the said sample ions upon the collision surface, back towards the electrostatic trap.

> 22. The method of claim 1, further comprising deflecting the sample ions between their exit from the storage device and their entrance to the electrostatic trap such that the said ions do not travel along a line of sight between the storage device exit and the electrostatic trap entrance.

> 23. The method of claim 1, further comprising: applying time dependent voltages to the trapping electrodes of the electrostatic trap so as to increase the electric field within the electrostatic trap until the arrival of the FWHM of the highest m/z to be analysed.

> 24. The method of claim 1, in which the electrostatic trap is of the orbitrap type having a central electrode and an outer electrode split into two sections, the method further comprising ramping to an electrostatic trapping potential on the said central electrode, and detecting an image current induced by the said trapped ions in the said split outer electrode.

> 25. The method of claim 24, in which the ions are confined in the axial direction and constrained to move around the central electrode by a hyper-logarithmic field having a potential distribution of the form: where:

> r and z are cylindrical coordinate, z=0 being the plane of symmetry of the field; C, k, Rm(>0) are constants, and k>0 for positive ions.

26. The method of claim 24, further comprising compen-65 sating the field within the orbitrap so as to minimise field perturbation within a volume occupied by the sample ions therein.

27. The method of claim 26, in which the field is compensated by applying a time dependent voltage to a field compensator in the orbitrap.

28. The method of claim 1, further comprising determining the optimum duration of trapping within the said storage 5 device.

29. The method of claim 1, further comprising filtering out a proportion of the sample ions prior to their introduction into the storage device, in accordance with their m/z ratio. **30**. A mass spectrometer comprising:

- (a) an ion source arranged to supply a plurality of sample ions to be analysed, each of which has a mass-to-charge ratio m/z;
- (b) an ion storage device comprising a plurality of storage device poles and having a storage device entrance end 15 through which the said sample ions are received and a storage device exit end through which the said sample ions may exit;
- (c) a voltage source arranged to supply a trapping voltage to the storage device poles so as to contain at least a 20 proportion of the sample ions received through the storage device entrance end of the storage device within a volume ρ of the storage device in a trapping mode during at least a part of a trapping period, the thus trapped ions each having a kinetic energy E_k such that 25 there is an average kinetic energy \overline{E}_k of the ions in the volume p during the trapping period or the said part thereof, and to supply a release voltage to the storage device in an ion ejection mode so as to controllably release at least some of the said sample ions contained 30 within the said volume p of the storage device through the storage device exit end, the release voltage being of a magnitude such that the potential difference then experienced by the ions across the volume ρ is greater than the said average kinetic energy \overline{E}_k during the 35 trapping period or said part thereof, and further wherein the release voltage is such that the electric field generated thereby at any first point across the volume p, upon application of the said release voltage, is no more than 50% greater or smaller than the electric field $_{40}$ generated thereby at any other second point across the volume ρ ; and
- (d) an electrostatic trap having an electrostatic trap entrance arranged to receive those ions released through the storage device exit end and meeting the 45 criteria imposed by the applied trapping and release potentials, as a convolution of bunched time of flight distributions for each m/z, each distribution having a full width at half maximum (FWHM); the electrostatic trap further comprising a plurality of electrodes 50 arranged to trap ions received through the electrostatic trap entrance therebetween so that the said trapped ions describe movement having periodic oscillations in at least one direction.

storage device poles are each segmented into two or more axially separate segments, the voltage source being arranged to supply a differential DC voltage between a one of the axially separated segments and the other or another of the axially separated segments. 60

32. The mass spectrometer of claim 31, in which the segments of the storage device poles are radially spaced from one another to define a trapping volume including the volume of therebetween, the axial length of the segment of each end pole most adjacent to the storage device exit being 65 static trap, the method comprising: no greater than twice the inscribed diameter defined by the radial spacing of those end pole segments.

33. The mass spectrometer of claim 32, in which the storage device further comprises an end cap defining therein a storage device exit, the axial distance between the end cap and a point mid-way along the end pole segments being no less than the said inscribed diameter.

34. The mass spectrometer of claim 30, further comprising a lens arrangement between the storage device exit and the electrostatic trap entrance, the lens arrangement being arranged to deflect the path of ions between the storage device and electrostatic trap such that they do not travel along a direct line of sight between the interior of the said storage device and the interior of the said electrostatic trap.

35. The mass spectrometer of claim 34, in which the lens arrangement is arranged to focus the ions into the electrostatic trap at an angle tangential to a central plane thereof.

36. The mass spectrometer of claim 30, in which the electrostatic trap is an orbitrap comprising a first, central electrode and a second, outer electrode, the second, outer electrode being split into two sections, the mass spectrometer further comprising means for detecting ions constrained within the orbitrap.

37. The mass spectrometer of claim 36, in which the voltage source is further arranged to supply a potential to the said central electrode, the mass spectrometer further comprising a voltage supply controller which is arranged to control the ramping of the potential on the central electrode in the electrostatic trap so that it occurs over a period of time which encompasses the duration of application of the release voltage to the storage device to cause the said controlled ejection of ions therefrom.

38. The mass spectrometer of claim 37, in which the voltage supply controller is arranged to apply the said release voltage to the storage device so that the ions arrive at the electrostatic trap whilst the potential on the central electrode is between (D1/D2)^{1/2} V and V, where V is the final static voltage applied to the central electrode, D1, is the outer diameter of that central electrode, and D2 is the inner diameter thereof.

39. The mass spectrometer of claim 30, further comprising field compensation means, the voltage supply being further arranged to supply a compensation voltage to the field compensation means so as to minimize field perturbation within the electrostatic trap.

40. The mass spectrometer of claim 30, further comprising a collision surface located downstream of the electrostatic trap, the ions exiting the storage device being focussed in time of flight onto the said collision surface so as to generate fragment ions for capture by the electrostatic trap.

41. The mass spectrometer of claim 30, further comprising ion cooling means arranged upstream of the said storage device.

42. The mass spectrometer of claim 30, further comprising a mass filter arranged upstream of the said storage device.

43. The mass spectrometer of claim 30, in which the 31. The mass spectrometer of claim 30, in which the 55 storage device is arranged to receive ions substantially along a first longitudinal axis, and to release the said ions along a second, substantially orthogonal axis.

> 44. The mass spectrometer of claim 43, in which the storage device is curved.

> 45. The mass spectrometer of claim 30, further comprising an ion deflector downstream of the storage device.

46. The mass spectrometer of claim 30, further comprising an ion energy booster upstream of the electrostatic trap.

47. A method for injecting sample ions into an electro-

supplying a trapping voltage to an ion storage device to trap at least some of a plurality of sample ions within

the storage device during at least a part of a trapping period, wherein each sample ion has a mass-to-charge ratio m/z;

- supplying a release voltage to the storage device, wherein the release voltage is of a magnitude such that the ⁵ sample ions released from the storage device arrive at an electrostatic trap as a convolution of bunched time of flight distributions for each m/z;
- receiving the sample ions released from the storage device through an entrance of the electrostatic trap, the sample¹⁰ ions arriving as a convolution of bunched time of flight distributions for each m/z, each distribution having a full width at half maximum (FWHM); and
- trapping the received sample ions within the electrostatic trap by applying a potential to a plurality of electrodes within the electrostatic trap such that the sample ions describe movement having periodic oscillations in at least one direction.

48. The method of claim **47** wherein supplying a release voltage to the storage device includes ensuring that an electric field experienced by the ions upon leaving the storage device is approximately uniform and ensuring that a potential drop experienced by each ion upon release from the storage device is larger than an average kinetic energy of the ions when trapped in the storage device.

49. A mass spectrometer comprising:

- an ion source configured to supply a plurality of sample ions to be analysed, each of which has a mass-to-charge ratio m/z;
- an ion storage device comprising a plurality of storage device poles and having a storage device entrance through which the sample ions are received from the ion source and a storage device exit through which the sample ions are released;

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- a voltage source configured to supply a trapping voltage to the storage device poles so as to contain at least a portion of the sample ions received through the storage device entrance within a volume ρ of the storage device in a trapping mode during at least a part of a trapping period, and to supply a release voltage to the storage device to controllably release at least some of the sample ions contained within the volume ρ through the storage device exit, wherein the release voltage is of a magnitude such that the sample ions released from the storage device arrive as a convolution of bunched time of flight distributions for each m/z at an electrostatic trap; and
- the electrostatic trap having an electrostatic trap entrance arranged to receive the ions released through the storage device exit as a convolution of bunched time of flight distributions for each m/z, each distribution having a full width at half maximum (FWHM), the electrostatic trap further comprising a plurality of electrodes arranged to trap ions received through the electrostatic trap entrance therebetween so that the trapped ions describe movement having periodic oscillations in at least one direction.

50. The mass spectrometer of claim **49** wherein the voltage source is configured to supply the release voltage to ensure that an electric field experienced by the ions upon leaving the storage device is approximately uniform and to ensure that a potential drop experienced by each ion upon release from the storage device is larger than an average thermal energy of the ions when trapped in the storage device.

* * * * *



United States Patent [19]

Makarov

[54] MASS SPECTROMETER

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[56] References Cited

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U.S. PATENT DOCUMENTS

4,982,088	1/1991	Weitekamp	250/291
5,528,031	6/1996	Franzen	250/291

5,886,346

Mar. 23, 1999

OTHER PUBLICATIONS

Blauth, E.W.: "Dynamic mass spectrometers", Elsevier Publishing Co., Amsterdam, 1966, 117–121.

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[57] ABSTRACT

A mass spectrometer comprises an ion source (11), an ion injection arrangement (12), field generator defined by shaped electrodes (14, 16) and a detector (18) to detect ions. The electrodes (14, 16) are shaped so as to provide therebetween a field of substantially hyper-logarithmic form whereby ions can be trapped within a potential well of the field for analysis.

22 Claims, 5 Drawing Sheets





FIG. 1.



FIG. 2.











FIG. 6.







FIG. 8.



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MASS SPECTROMETER

BACKGROUND OF THE INVENTION

1. Field of the Invention

This invention relates to improvements in or relating to a mass spectrometer and is more particularly concerned with a form of mass spectrometer which utilizes trapping of the ions to be analyzed.

2. Description of the Prior Art

Molecular or atomic weight of a substance is a useful 10 characteristic which, if detected, can enable the substance to be identified. A mass spectrometer is a measuring instrument which can determine the molecular weight of a substance or other molecule introduced into it for analysis. Mass Spectrometers operate in a number of different ways, however the 15 present invention is concerned particularly with mass spectrometers in which ions are trapped or confined within a particular region of space for analysis purposes. Known types of mass spectrometers of this type are the so-called "quadrupole ion trap" spectrometers and "ion cyclotron 20 resonance" spectrometers.

Quadrupole ion trap mass spectrometers currently available use a three-dimensional quadrupole electric field which oscillates at radio frequencies to trap ions. The ions can then be ejected from the field selectively on the basis of mass/ 25 spectrometer according to the present invention; charge ratio enabling the device to operate as a mass spectrometer. This form of spectrometer can be produced relatively inexpensively and relatively small in size, making it a popular choice as a mass selective detector for gas chromatographs (GC-MS).

Ion cyclotron resonance (ICR) mass spectrometers currently available use a combination of an electric field and a very strong magnetic field to trap ions. The trapped ions spiral around the magnetic field lines with a frequency related to the mass of the ion. The ions are then excited such 35 that the radii of their spiralling motion increases and as the radii increase the ions are arranged to pass close to a detector plate in which they induce image currents. The measured signal on these detector plates as a function of time is related Conventional techniques such as Fourier transformation can be applied to the measured signal to obtain the component frequencies of the ions and hence produce a frequency (and hence mass) spectrum. This type of mass spectrometer is 45 able to produce a very high degree of mass resolution.

However, there-are disadvantages associated with the known forms of mass spectrometer described above. For instance, while the quadrupole ion trap mass spectrometer can be constructed small and cheaply, the mass resolution 50 and mass range obtained is not very high unless the analysis is carried out using very slow scanning. While this is adequate for gas chromatograph mass measurement, it limits the applicability to molecular weight molecules of a biochemical nature. Furthermore, with the ion cyclotron resonance mass spectrometer described above, in order to pro-55 vide the high magnetic field necessary for the spectrometer to work efficiently, it is necessary to provide a super conducting magnet which in itself is very expensive. Furthermore, a super conducting magnet of the type necessary requires, with technology currently available, the use of 60 liquid helium to cool it and as a continuous supply of this is required, it necessarily results in high running costs of the spectrometer due to the relatively high cost of liquid helium.

SUMMARY OF THE INVENTION

It is an object of the present invention to provide an improved mass spectrometer in which the high mass resolution associated with the ion cyclotron resonance mass spectrometer can be achieved in a small and relatively inexpensive mass spectrometer.

According to the present invention therefore there is provided a mass spectrometer comprising an ion source to produce ions to be analyzed, electric field generation means to produce an electric field within which said ions can be trapped and detection means to detect ions according to their mass/charge ratio wherein said electric field defines a potential well along an axis thereof and said ions are caused to be trapped within said potential well and to perform harmonic oscillations within said well along said axis, said ions having rotational motion in a plane substantially orthogonal to said axis.

Preferably said electric field produced by the electric field generation means is of substantially "hyper-logarithmic form".

With this arrangement it is possible to detect ion mass/ charge ratio with a high degree of resolution in a simple and inexpensive manner.

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is a schematic side view of one form of mass

FIG. 2 is a side view to a larger scale of a part of FIG. 1 showing the field generation arrangement and measurement chamber;

FIG. 3 shows a schematic view of a part of FIG. 1 to a larger scale showing part of one form of ion injection arrangement:

FIG. 4 shows a graphical representation of one form of the potential distribution of the electric field provided by the field generation arrangement;

FIG. 5 shows a diagrammatic representation of the movement of trapped ions in the measuring chamber with the electric field of FIG. 4;

FIG. 6 shows a diagrammatic representation of the moveto the number and frequencies (hence mass) of the ions. 40 ment of ions from the ion injection arrangement to the measuring chamber;

> FIG. 7 shows a side view similar, to FIG. 2 illustrating the movement of the ions in a measurement chamber in the axial direction after excitation;

> FIG. 8 shows a diagrammatic representation, partly in section, of one form of ion ejector from the measurement chamber in the MSI mode of operation; and

> FIG. 9 shows graphical representations of various parameters of a mass spectrometer indicating the performance of the mass spectrometer of the present invention (1) and similar parameters of a conventional ICR mass spectrometer.

DETAILED DESCRIPTION OF A PREFERRED EXEMPLARY EMBODIMENT

Referring now to FIG. 1, there is shown a schematic representation of a mass spectrometer 10 which comprises an ion source 11, ion injection arrangement 12, field generator means 13 defined by the outer and inner shaped electrodes 14, 16 which define between them a measurement cavity 17 and one or more detectors 18 to detect the ions, either trapped in the field or ejected therefrom in a manner to be hereinafter defined

The ion source 11 comprises either a continuous or pulsed ion-source of conventional type and produces an ion stream which exits through a slit 19 in a front part thereof.

The ion injection arrangement 12 (shown more clearly in FIG. 3) comprises two concentric cylinder electrodes 21, 22, the outer electrode 21 being of substantially larger diameter than the inner electrode 22. The outer cylinder electrode 21 has a tangential hole through which ions from the source pass into the region between the outer and inner electrodes 21, 22. The injection arrangement 12 is mounted round the field generator means and is in connection therewith in a manner which will be described hereinafter. The outer cylindrical electrode 21 is stepped at ends thereof for a 10 segments, if desired. reason which will become hereinafter apparent. While in the embodiment described, the inner cylindrical electrode 22 is formed as a separate electrode, it is possible to use a top surface 36 of the shaped electrode 16 as indicated in FIG. 1

The field generation arrangement 13 is disposed within the confines of inner cylinder electrode 22 and includes two shaped electrodes, internal and external field generator electrodes 14, 16 respectively. The space 17 between the internal and external shaped electrodes 14, 16 forms the measure- 20 ment chamber. The electrodes 14, 16 are shaped for a reason which will become hereinafter apparent. The outer shaped electrode 16 is split into two parts 23, 24 by a circumferential gap 26, an excitation electrode part 23 and a detection electrode part 24. The circumferential gap 26 between the ²⁵ outer electrode parts 23, 24 allows ions to pass from the injection arrangement to the measurement chamber 17 in a manner to be hereinafter defined.

The cylindrical and shaped electrodes are connected to 30 respective fixed voltage supplies via a potential divider arrangement 27 which allows a desired voltage to be applied to the electrodes.

The measurement chamber 17 is linked to a vacuum pump UHV of approximately 10⁻⁸ Torr or lower.

The internal and external shaped electrodes 14, 16 when supplied with a voltage will produce respective electric fields which will interact to produce within the measurement chamber 17 a so-called "hyper-logarithmic field". The $_{40}$ potential distribution of a hyper-logarithmic field, is shown in FIG. 4 and is described in cylindrical coordinates (r,z) by the following equation:

$u(r,z)=k/2((z-a)^2-r^2/2) +bln(r/c)+d$

where a, b, c, d and k are constants. It can be seen from this figure that such a field has a potential well along the axial (Z) direction which allows an ion to be trapped within such potential well if it has not enough energy to escape. The field is arranged such that the bottom of the potential in the radial 50 direction (i.e. along axis r in FIG. 4) lies along the longitudinal axis of the measurement chamber 17 shown in FIGS. 1 and 2. While for the purposes of illustration of the present invention a hyper-logarithmic field will be described, it is thought that other forms of field will be capable of being 55 used, the only restriction on the form of field generated being that the field defines in potential terms a threedimensional well in which ions can be trapped, and ions are prevented from striking an inner electrode by virtue of rotational motion about this electrode. 60

A suitable detector which may be connected to a microprocessor based circuit is provided which analyzes the signal in accordance with conventional Fourier analysis techniques by detecting one or more of the following frequency characteristics of the ions in the chamber 17, i.e. harmonic 65 motion in its axial direction, oscillation in the radial direction and the frequency of angular rotation. The most appro-

priate frequency to give the required high performance is the harmonic motion in the axial direction. These frequencies can be detected while the ions are in the measurement chamber 17. The ions may also be detected after they have been ejected from the chamber 17, as desired or as appropriate. Where detection in the measurement chamber 17 is used, it is possible to use one half of the outer electrode 16 as a detector as will be described hereinafter. Each of the electrodes 14, 16 may be split into two or more electrode

In use, ions to be measured are produced by the ion source 11, focused and accelerated by plates 27-31 and leave the ion source 11 through entrance slit 19.

The ion source 11 is directed towards a tangential inlet to form entirely the function as inner cylinder electrode 22. 15 aperture (not shown) in the outer cylindrical electrode 21 and the ions enter the injection cavity 32 between the cylindrical electrodes 21, 22 with a small axial velocity component so that the ions move axially away from the inlet. The field produced between the two cylindrical electrodes 21, 22 causes the ions to enter a spiral trajectory around the inner cylindrical electrode 22.

In order to inject the ions from the injection arrangement 12 into the measurement cavity 17, it is necessary to modify the electric field produced by the cylinder electrodes 21, 22 (and 36 where appropriate) to define a potential valley which is directed towards the circumferential gap 26 between the excitation and detection electrode parts 23, 24. In the apparatus of the present invention, this is achieved by providing steps in the cylinder electrode walls 25 which, in combination with the fringing effects caused by the circumferential gap modifies the field in the manner desired. Of course, it may be possible to achieve the same effect using different means as desired or as-appropriate. By increasing the voltage applied to the electrodes 21, 22, 23, 24 with time, which operates to evacuate the measurement chamber to a $_{35}$ the sides of the potential well are increased in gradient thereby forcing the ions to oscillate within the confines of this valley. Furthermore, as the voltage increases, the field intensity increases and therefore the force on the ions towards the longitudinal axis increases thus decreasing the radius of spiral of the ions. Thus it can be seen that the ions converge into the gap 26 by virtue of being forced to rotate in spirals of smaller radius and by a potential well caused by modification of the field produced by the electrodes 21, 22, 23, 24. This is shown schematically in FIG. 6. Of course, the 45 injection arrangement 12 can take any form as desired or as appropriate, for example electrodes 21, 22 need not be present and electrodes 23, 24 can be segmented, and a part of the field can be switched off during injection and switched on again to trap the ions once injection has been completed. The present arrangement has been developed to provide greater sensitivity.

> After sufficient ions have been directed into the measurement chamber 17, the voltage supply to spaced electrodes 14, 17 can be maintained constant and the voltage supply to the cylinder electrodes 21, 22 can be changed such that all ions outside the hyper-logarithmic field are lost in the injection arrangement 12.

> The shaped electrodes 14, 16 in the field generation arrangement are shaped so as to have the shape of equipotential surfaces in, the required potential distribution. The hyper-logarithmic field is created in the measurement chamber 17 by the electrodes 14, 16 and the ions injected from the injection arrangement 12 through gap 26 are maintained within the potential well in this field so as not to strike inner electrode 14 by ensuring that they have sufficient rotational energy to orbit the electrode 14 in a spiral trajectory. Thus the ions to be analyzed are trapped in the field and are forced

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to oscillate back and forth within the confines of the well created by the hyper-logarithmic field in a spiral trajectory around the central electrode 14.

Once the ions are trapped in the hyper-logarithmic field, various methods of analysis can be used as are described hereinafter.

After mass analysis has been completed, any remaining ions in the injection or measuring chamber are swept away by, changing the voltage supply to the electrodes 14, 16 for a short time.

Mass analysis can be carried out using the mass spectrometer of the invention in one of two modes which will be considered in turn:

1. Fourier Transform Mode

There are three characteristic frequencies of oscillation within the field. The first is the harmonic motion of the ions in the axial direction where they oscillate in the potential well with a frequency independent of energy in this direction.

The second characteristic frequency is oscillation in the radial direction since not all the trajectories will be perfectly circular.

The third frequency characteristic of the trapped ions is 25 the frequency of angular rotation.

In order to detect the frequencies of oscillations the motion needs to be coherent. The radial and rotational oscillations are not coherent since ions are injected into the measurement cavity 17 continuously over a period of time, 30 and hence the distribution of ions around the inner shaped electrode 14 is random. It is easiest to induce coherence in the axial oscillations and therefore the outer electrode 16 is formed in two parts 23, 24 as described above for this purpose. If a voltage pulse is applied to one part 23 of this 35 electrode, the ions which exist as a disc in the measurement chamber 17 after passing through the gap 26 between the two parts 23, 24, will receive a force toward the other part 23 or 24 in the axial direction. After this pulse the voltages on the two parts 23, 24 can once again be made equal and 40the ions will then oscillate with harmonic motion in the potential well of the field in the axial direction. One or both parts 23, 24 of the outer shaped electrode 16 is then used to detect image current as the ions oscillate back and forward. The Fourier Transform of the signal from the time domain 45 perturbations. Again high resolutions are only possible at the to the frequency domain can thus produce a mass spectrum in conventional manner. It is in this mode of detection with which high mass resolutions are possible.

2. The Mass-Selective Instability (MSI) Mode

The second mode of mass detection involves ejection of the ions from the potential well in the hyper-logarithmic field and collection on a detector.

This mode of operation is analogous to that used in $_{55}$ conventional quadrupole ion traps, but differs greatly in that in this device there is no instability in the radical direction.

Although the principal analysis method used in terms of utilising the important advantages of the present invention would be the Fourier Transform mode, there are certain instances where the MSI mode is useful. For example one mass, can be stored for subsequent MS/MS analysis, by ejecting all other masses from the trap, or high intensity signals from unwanted components can be ejected to improve dynamic range.

In this method, the voltage applied to the electrodes 14, 16 is varied sinusoidally with time as in a quadrupole or

quadrupole ion trap device, giving two, possible regimes of mass instability.

a) Parametric Resonance

If the voltage between the inner and outer shaped electrode 14, 16 of the spectrometer is varied sinusoidally, then the equations describing ion motion within the trap are the well-known Mathieu equations. In a complete analogy with the quadrupole or quadrupole ion trap, the solutions of the equations of motion can be expressed in terms of two parameters a and q, and can be represented graphically on a stability diagram.

Application of the appropriate frequency for a given mass results in excitation of oscillations in the axial direction, and after sufficient excitation results in ejection from the measurement chamber 17. A convenient means of detection of the ions is collision with a conversion dynode 32 in the outer electrode 16 which generates secondary electrons which can be accelerated away to a detector (FIG. 8). The main advantage over the quadrupole ion trap is that the magnitude of the radio frequency voltages required are much lower, which means that the mass range of the spectrometer in this mode is effectively unlimited. The mass range of the quadrupole ion trap in conventional scan mode is limited in practice to a few thousand Daltons as very high voltages (>10,000) are required at high mass whereas only a few tens of volts are required in the spectrometer of the present invention.

With this method there are two types of scanning with regard to mass resolution. The first is a rapid scan mode which provides around unit mass resolution. The second regime utilizes the addition of some anharmonic field perturbations which allow the achievement of very high resolutions but at the expense of scan speed. The slower the scan speed the higher the resolution.

b) Resonant Excitation

In this mode of operation the sinusoidal oscillations are applied to one half 23, 24 of the outer shaped electrode 16 at the resonant axial frequency of a particular mass. As above, both low and high resolution modes of operation are possible. The disadvantage of this mode at low resolution compared to the parametric excitation mode is the presence of a number of side resonances which leads to artefacts. However the resonant excitation mode becomes competitive with the parametric excitation mode at high resolution modes of scanning which make use of anharmonic field expense of scan speed. Whether parametric or resonance excitation is the best MSI mode for high resolution depends on the application in which it is to be used. For example parametric resonance does not show a large dependence on 50 beam width, but resonant excitation provides higher scanning rates at the high resolution due to a faster rate of energy acquisition during excitation.

The main advantage of the spectrometer of the present invention over the prior art type of spectrometers, and in particular the ion Cyclotron Resonance (ICR) specification, is much better detection efficiency at high mass. This arises due to the fact that the signal to noise ratio (S/N) is proportional to the image current frequency in an ICR spectrometer the frequency of oscillation decreases as I/M (M being the mass to charge ratio of the ion). With the spectrometer of the present invention the frequency of oscillation decreases as I/M1/2 and hence decreases much more slowly. Thus the spectrometer of the present invention should realise a 30-100 increase in detection efficiency in the 10-100 k Da range. This high mass capability is important in the application of mass spectrometers to biological compounds.

Comparatively the spectrometer of the present invention has less mass resolution at low masses (>1000) than the ICR specification. This arises due to the higher field accuracy in the ICR spectrometer.

Furthermore, the space charge effects (related to the number of ions and hence dynamic range) which can be tolerated in the spectrometer of the present invention is greater than can be tolerated in an ICR spectrometer. This arises due to the fact that the ions are distributed along a longer trajectory and there is some shielding of the ions from 10 allowing the ions to be directed into said field generation each other due to the presence of the central electrode.

These comparisons are illustrated graphically in FIG. 9 of the drawings.

It will be appreciated that with the arrangement of the present invention, it is possible to provide a mass spectrom- 15 eter which is relatively simple and inexpensive to produce which allows high resolution measurements to be made.

It is of course to be understood that the invention is not intended to be restricted to the details of the above embodiment which is described by way of example only.

What is claimed is:

1. A mass spectrometer comprising an ion source to produce ions to be analyzed, electric field generation means to produce an electric field within which said ions can be trapped and detection means to detect ions according to their 25 mass/charge ratio wherein said electric field defines a potential well along an axis thereof and said ions are caused to be trapped within said potential well and to perform substantially harmonic oscillations within said well along said axis, said ions having rotational motion in a plane substantially 30 orthogonal to said axis.

2. A mass spectrometer according to claim 1 wherein the electric field generated is of substantially hyper-logarithmic form and is defined by the following equation:

 $U(r,z)=k/2[(z-a)^2-r^2/2]+bln(r/c)+d$

where r, z are cylindrical coordinates and a, b, c, d, k are constants with c>O and b, k>O.

3. A mass spectrometer according to claim 2 wherein the field generation means comprises a pair of electrodes having a shape defined by the equations $z_1(r)$ and $z_2(r)$ respectively and a potential defined by the equations $U(r,z_1(r))=U_1$ and $U(r,z_2(r))=U_2$.

4. A mass spectrometer according to claim 3 wherein said 45 electrodes are coaxial, one electrode forming an outer electrode and another forming an inner electrode.

5. A mass spectrometer according to claim 3 wherein at least one of said electrodes comprises from at least two sections positioned adjacent each other with a gap therebetween.

6. A mass spectrometer according to claim 1 in which an ion injection arrangement is provided which generates an injection electric field which injects ions into the electric field produced by the field generation means to be trapped therein.

7. A mass spectrometer according to claim 6 wherein the ion injection arrangement comprises electrodes disposed externally of the field generation means so as to surround at least a part of the Field generation means.

8. A mass spectrometer according to claim 7 wherein said ion injection arrangement comprises a pair of coaxial cylinder electrodes.

9. A mass spectrometer according to claim 7 wherein at least one of said electrodes is adapted to modify the injection electric field to produce a potential well into which ions can pass so as to be is directed into the electric field produced by the field generation means to be trapped therein.

10. A mass spectrometer according to claim 9 wherein after passage into the potential well in the injection fields, a voltage applied to the electrodes is varied to reduce the magnitude of oscillations of the ions within the well thereby means through a gap between said electrodes.

11. A mass spectrometer according to claim 7 wherein the injection electric field produced causes ions to follow a spiral trajectory around an inner of said electrodes.

12. A mass spectrometer according to claim 6 wherein said ion source includes acceleration and focusing means to accelerate and focus said ions into said ion injection arrangement.

13. A mass spectrometer according to claim 12 wherein 20 said acceleration and focusing means comprises a plurality of charged plates.

14. A mass spectrometer according to claim 12 wherein after passing through said acceleration and focusing means, ions are directed through a tubular member.

15. A mass spectrometer according to claim 1 wherein the harmonic oscillations of said ions are excited by variation of a voltage applied to said field generation means.

16. A mass spectrometer according to claim 1 wherein said detection means acts to detect said ions by detection of an image current induced on a part of said electrodes.

17. A mass spectrometer according to claim 1 wherein said ions are excited and ejected from said field for detection.

18. A mass spectrometer according to claim 17 when said 35 detection means detects secondary particles produced by collision of ions with at least a portion of said detection means.

19. A mass spectrometer according to claim 18 wherein said detection means comprises a dynode and a secondary electron detector, said ions after being arranged to collide with said dynode thereby to produce secondary electrons, said secondary electrons being detected by said detector.

20. A mass spectrometer according to claim 1 further including fragmentation means which is operable to split said ions produced by said ion source into smaller ions thereby allowing the spectrometer to operate in MS/MS configuration.

21. A mass spectrometer according to claim 20 wherein said fragmentation means is operable to fragment selected 50 said ions when trapped in said electric field, non-selected ions being ejected from said field.

22. A mass spectrometer according to claim 1 further comprising at least one of electrodes having at least two sections positioned adjacent each other with a gap there 55 between, and an ion injection arrangement which is provided to generate an injection electric field that injects ions into the electric field produced by the field generation means to be trapped therein, and wherein said ion injection arrangement is operable to inject ions into the field produced by said 60 field generation means through said gap in said electrodes.